

George R. Clark,* Deborah M. Tonei, Warren R. Roper and L. James Wright

Department of Chemistry, The University of Auckland, Private Bag 92019, Auckland, New Zealand

Correspondence e-mail:
g.clark@auckland.ac.nz

Key indicators

Single-crystal X-ray study
 $T = 85\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.048
 wR factor = 0.120
Data-to-parameter ratio = 10.8

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3,3,6,6-Tetraphenyl-*trans*-tricyclo[3.1.0.0^{2,4}]hexane

The title compound, $\text{C}_{30}\text{H}_{24}$, was produced serendipitously whilst investigating the reactions of 3,3-diphenylcyclopropane with low-valent metal substrates. There are only a small number of crystal structure determinations of products from metal-catalysed (2 + 2)-cycloadditions of cyclopropenes. The product here is the centrosymmetric *trans* isomer.

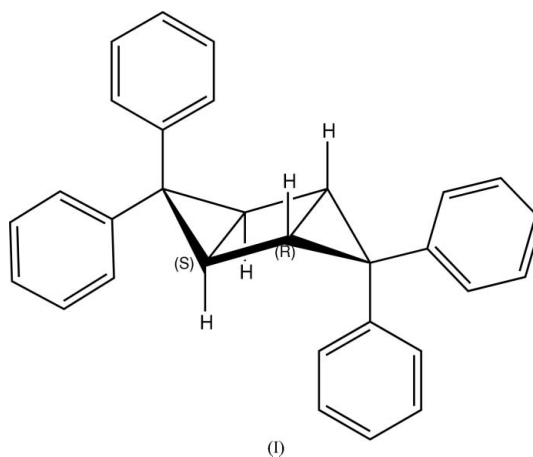
Received 9 December 2005

Accepted 3 January 2006

Online 11 January 2006

Comment

While exploring the interaction between $\text{Os}(\text{CO})_2(\text{PPh}_3)_3$ (Collins *et al.*, 1982) and 3,3-diphenylcyclopropene (Huang *et al.*, 2003) we observed the crystallization, in very low yield and along with several other products (all osmium organometallic compounds), of the [2 + 2]-cycloaddimerized product 3,3,6,6-tetraphenyl-*trans*-tricyclo[3.1.0.0^{2,4}]hexane. Cyclodimerization of cyclopropenes has previously been reported either directly by photolysis (Klimova *et al.*, 2000) or through catalysis by low oxidation state metal complexes, *e.g.* with nickel catalysts (Binger & Doyle, 1978) and with copper catalysts (Baird *et al.*, 1987). The molecule possesses a crystallographic centre of symmetry. The principal bond lengths and angles are listed in Table 1. All values are consistent with those previously obtained for other tricyclohexane structures where substitution is only on the apical C atom of the cyclopropane rings (Klimova *et al.*, 2000; Arrowood & Kass, 1999).



Experimental

Crystals of (I) were obtained as a very minor by-product of the reaction between $\text{Os}(\text{CO})_2(\text{PPh}_3)_3$ and 3,3-diphenylcyclopropene. We have not attempted to develop reproducible reaction conditions.

Crystal data

C₃₀H₂₄
M_r = 384.49
 Monoclinic, *P*2₁/*c*
a = 8.1918 (5) Å
b = 18.4683 (12) Å
c = 7.0314 (4) Å
 β = 98.709 (1)°
V = 1051.51 (11) Å³
Z = 2

D_x = 1.214 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 4253 reflections
 θ = 2.2–25.7°
 μ = 0.07 mm⁻¹
T = 85 (2) K
 Thick plate, colourless
 0.28 × 0.22 × 0.12 mm

Data collection

Siemens SMART CCD diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
*T*_{min} = 0.924, *T*_{max} = 0.970
 5914 measured reflections

1994 independent reflections
 1630 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.025
 θ_{max} = 25.7°
h = -9 → 9
k = -22 → 22
l = -8 → 8

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.048
wR (*F*²) = 0.120
S = 1.09
 1994 reflections
 184 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0494P)^2 + 0.4478P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.31 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

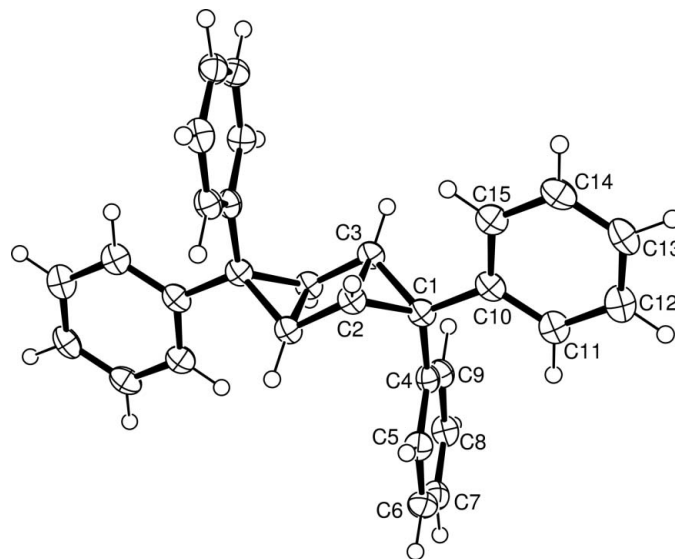


Figure 1

Structure showing 50% probability displacement ellipsoids for non-H atoms and H atoms as arbitrary spheres (Burnett & Johnson, 1996). [Symmetry code for unlabelled atoms: 1 - *x*, -*y*, -*z*.]

ORTEP^{III} (Burnett & Johnson, 1996); software used to prepare material for publication: SHELXTL (Siemens, 1995).

The authors thank FoRST for the award of an NZ S&T Post-doctoral Fellowship to DMT.

References

- Arrowood, T. L. & Kass, S. R. (1999). *Tetrahedron*, **55**, 6739–6748.
 Baird, M. S., Hussain, H. H. & Clegg, W. (1987). *J. Chem. Soc. Perkin Trans. 1*, pp. 1609–1612.
 Binger, P. & Doyle, M. J. (1978). *J. Organomet. Chem.* **162**, 195–207.
 Burnett, M. N. & Johnson, C. K. (1996). ORTEP^{III}. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
 Collins, T. J., Grundy, K. R. & Roper, W. R. (1982). *J. Organomet. Chem.* **231**, 161–172.
 Huang, J.-H., Lee, T.-Y., Swenson, D. C. & Messerle, L. (2003). *Inorg. Chim. Acta*, **345**, 209–215.
 Klimova, E. I., Garcia, M. M., Klimova, T., Toledano, C. A., Toscano, R. A. & Ramirez, L. R. (2000). *J. Organomet. Chem.* **605**, 89–95.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. Release 97-1. University of Göttingen, Germany.
 Siemens (1995). SHELXTL (Version 5), SMART (Version 4.050) and SAINT (Version 4.050). Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Table 1

Selected geometric parameters (Å, °).

C1–C4	1.504 (2)	C1–C2	1.523 (2)
C1–C3	1.514 (2)	C2–C3 ⁱ	1.496 (2)
C1–C10	1.517 (2)	C2–C3	1.540 (2)
C4–C1–C3	119.10 (14)	C3 ⁱ –C2–C1	112.10 (14)
C4–C1–C10	116.87 (14)	C3 ⁱ –C2–C3	90.32 (13)
C3–C1–C10	115.29 (14)	C1–C2–C3	59.23 (11)
C4–C1–C2	118.95 (14)	C2 ⁱ –C3–C1	111.89 (14)
C3–C1–C2	60.93 (11)	C2 ⁱ –C3–C2	89.68 (13)
C10–C1–C2	113.88 (14)	C1–C3–C2	59.83 (11)

Symmetry code: (i) -*x* + 1, -*y*, -*z*.

H atoms were located in a difference Fourier map and refined with individual isotropic displacement parameters [C–H = 0.946 (19)–1.027 (19) Å].

Data collection: SMART (Siemens, 1995); cell refinement: SAINT (Siemens, 1995); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: